## Answer Sheets for CHE654 Homework Set #5 (100 Points)

 $\underline{\text{Note}}$ : For all problems, submit a copy of your process flow diagram and a copy of your input summary of the process.

36. (20 points) Extracting Compound Y via	an Evaporator and a Stripper Using	g Compound X
Answer the following questions:		
Required heat transfer area of the evapora	$ator = \underline{\qquad} ft^2$	
Concentration of Compound <i>Y</i> in the liqu	or stream from the evaporator =	mol%
Concentration of Compound <i>Y</i> in the outl	et liquid stream of the stripper =	mol%
37. (20 points) Separation of Compound X,	II	
Answer the following questions:		
(a) Redundant parameters:		
(b) Column bottom flow rate =	lbmol/hr	
Distillate vapor flow rate =	lbmol/hr	
39. (20 points) Property Requirements and	PCES, II	
(a) Required parameters:		
(b) Values of estimated parameters:		
$MW = \underline{\hspace{1cm}}; T_C = \underline{\hspace{1cm}}$	$K; P_C = $ $N/m$	$1^2$

	$Z_{\rm C} = $ ; CPIG at 30	0 K =	J/kmol-K	
	DHFORM = J/kr	mol; DGFORM =	J/kmol	
	OMEGA =; DH	IVLB =	J/kmol	
	$VB = \underline{\hspace{1cm}} m^3/kmol$			
(c)	$H_V^{IG} =$	_ Btu/lbmol		
	$H_L = $	Btu/lbmol		
(d)	Tdew at 1 atm =	$^{\circ}\mathrm{C}$		
	Tdew at 50 atm =	$^{\circ}\mathrm{C}$		
41. (20 points) Recovering Acetone from a Wastewater Stream				
Answer the following questions:				
D:F molar ratio in Column 1 =				
Purity of acetone product stream = mole%				
42. (20 points) Extractive Distillation, II				
Answer the following questions:				
Mole % purity of butene in the distillate of the extractive column =				

Mole % purity of butadiene in the distillate of the solvent recovery column = \_\_\_\_\_